Saturated Heterocycles. 136 [1]. Synthesis of 2-Substituted 2-Azapyracridone Derivatives

Imre Huber, Ferenc Fülöp and Gábor Bernáth*

Institute of Pharmaceutical Chemistry, University Medical School, H-6701 Szeged, P. O. Box 121, Hungary

István Hermecz*

Chinoin Pharmaceutical and Chemical Works, Research Centre, H-1325 Budapest, P. O. Box 110, Hungary Received March 16, 1987

By means of substitution, addition and condensation on the secondary nitrogen of 9-methyl-1,2,3,4-tetra-hydro-2-azapyracridone (6), twenty new azapyracridones were prepared.

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In a previous paper [2] we reported the synthesis of new dipyrido[1,2-a:4,3-d]pyrimidin-11-one (2-azapyracridone) derivatives. For several of these compounds a catalytic intermolecular hydrogen-transfer reaction was observed [3].

Many papers currently deal with the examination of pyrido[2,1-b]quinazolin-11-ones (pyracridones [4]), which have promising pharmacological properties (Figure 1). For instance, the 8-isopropylpyrido[2,1-b]quinazoline-2-carboxylic acid (1, $R^1 = COOH$ and $R^2 = isopropyl$) exhibits a valuable anti-asthmatic effect [5a,b]. Some of the ring C homologues of pyracridone 2a with favourable pharmacological effects have been isolated also from plants [6a-e]. The ring A homologues of pyracridones 2b were recently synthesized [7a-d] for pharmacological [8a,b] and structural [9a,b] investigations.

Figure 1

Compounds 3, 4, 5 and 6 with additional hetero atoms in ring A or C can be prepared analogously to the syntheses of 1 and 2, but from heterocyclic starting materials (e.g. heterocyclic β -ketoesters, nicotinates, 2-aminopyridazines, etc.) (Figure 2). For the recently synthesized

thieno[3,4-d]pyrido[1,2-a]pyrimidin-10-one-7-carboxylic acid (3), the starting material was a sulphur-containing β -ketoester [10]. Compound 3 and 11-oxopyridazino[3,2-b]-quinazoline-8-carboxylic acid (5) [11] have favourable antiallergic and anti-asthmatic effects. Ring C homologues of compounds 4 have various pharmacological properties [12a-d].

Figure 2

As a continuation of our synthetic work on bridgehead nitrogen-containing heterocycles [7a,9a,b], we now report some chemical transformation of 2-azapyracridones. 9-Methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (6) was prepared in 95% yield by a modification of the earlier route [3]. The ring-closure reaction of 3-methoxycarbonyl-4-piperidone hydrochloride and 2-amino-6-methylpyridine was carried out in polyphosphoric acid. The reactivity of the secondary nitrogen in position 2 was examined in the reactions shown in the Scheme.

The methyl acrylate and acrylonitrile additions of 6 gave compounds 7a and 7b, according to anti-Markovni-kov orientation (Table). Ester 7a was reduced with lithium aluminum hydride to the 1,3-aminoalcohol 7c; from 7a

Scheme

with hydrazine hydrate, the hydrazide 7d formed. Compound 6 could be transformed with isocyanates and isothiocyanates to derivatives of urea and thiourea 8a-c even at room temperature. The secondary nitrogen reacted readily with alkyl halides in dry acetone in the presence of potassium carbonate, resulting in compounds 9a-g. Compounds 10a-d were formed by Schotten-Baumann acylation. Derivative 10e was prepared from 10d with N-methylpiperazine. A fast reaction was observed between 6 and dimethyl acetylenedicarboxylate (DMAD), while with cyclohexene oxide a slow addition took place to afford the trans-1,2-aminoalcohol 12.

The 60 MHz ¹H nmr spectra corresponded to the structures given in the Scheme.

EXPERIMENTAL

Melting points were determined on a Boetius micro melting point apparatus and are uncorrected. The ir spectra were recorded in potassium bromide discs pills on a Unicam SP 200 spectrometer. Physical and analytical data on the compounds prepared are listed in the Table.

9-Methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (6).

A mixture of 2-amino-6-methylpyridine (1.08 g, 10 mmoles) and 3-

methoxycarbonyl-4-piperidone hydrochloride (1.94 g, 10 mmoles) was heated in polyphosphoric acid (10 g, FLUKA) for 6 hours, with stirring in an oil-bath at 120°. The mixture was then cooled to about 70° and water (10 ml) was added. After neutralization with 10% sodium hydroxide solution, the product was extracted with a mixture of chloroform-2-propanol 3:1 (3 x 100 ml). The combined extract was dried (sodium sulfate) and evaporated to afford 1.98 g (95%) of 6, mp 135-136° (lit [2] yield 86%, mp 134-135°).

2-[$(\beta$ -Methoxycarbonyl)-ethyl]-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2- α :4,3- α]pyrimidin-11-one (7a).

A mixture of 6 (0.43 g, 2 mmoles) and methyl acrylate (0.17 g, 2 mmoles) in 20 ml of methanol was refluxed for 4 hours. After evaporation, the crude oily product 7a was crystallized from di-isopropyl ether.

 $2(\beta$ -Cyanoethyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]-pyrimidin-11-one (7b).

A mixture of 6 (0.43 g, 2 mmoles) and acrylonitrile (0.11 g, 2 mmoles) in 20 ml of methanol was refluxed for 3 hours. After removal of the solvent, the yellow crystalline residue 7b was recrystallized from methanol.

2- $(\gamma$ -Hydroxypropyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-a]pyrimidin-11-one (7c).

To a stirred suspension of lithium aluminum hydride (0.1 g) in 10 ml of dry THF, the ester 7a (0.3 g, 1 mmole) was added under ice cooling. After 5 minutes, the mixture was worked up with the routine procedure. The crude oily product was crystallized from ethyl acetate.

 $2\cdot(\beta-\text{Hydrazidoethyl})\cdot 9-\text{methyl-}1,2,3,4-\text{tetrahydro-}11H-\text{dipyrido}[1,2-a:4,3-d]$ pyrimidin-11-one (7d).

Table
Physical and Analytical Data of the 2-Aza-9-methyl-1,2,3,4tetrahydropyracridones 7-12

No.	Χ	n	R	Yield (%)	Mp (C ^o) solvent	Molecular formula	<i>Anal.</i> C(%)	(Calcd./ H(%)	Found) N(%)	IR v max (cm ⁻¹)
7a	CO ₂ CH ₃	-	-	95	69 - 70	C ₁₆ H ₁₉ N ₃ O ₃	63.77	6.36	13.94	1470, 1650,
					diisopropyl ether		64.03	6.86	14.39	1720
7b	CN	-	-	90	164 - 166	C ₁₅ H ₁₆ N ₄ O	67.14	6.01	20.88	1490, 1680,
					methanol		67.48	6.53	20.46	2300
7c [a]	CH ₂ OH	-	-	95	126 - 127	C ₁₅ H ₁₉ N ₅ O ₂	65.91	7.01	15.37	1470, 1650,
					ethyl acetate		66.20	7.27	15.59	3100
7d [b]	CONHNH ₂	-	-	60	175 - 177	C ₁₅ H ₁₉ N ₅ O ₂	59.79	6.35	23.24	1660, 3180,
					methanol		60.00	6.92	23.50	3310
8a	0	-	Ph	95	204 - 205	$C_{19}H_{18}N_4O_2$	68.25	5.43	16.76	1480, 1660,
					benzene		68.08	4.99	17.04	1620
8b	S	-	Ph	90	210 - 213	C ₁₉ H ₁₈ N ₄ OS	65.12	5.18	16.99	1480, 1660,
			CICI		methanol		64.94	5.51	16.73	1520
8c	S	-		90	228 - 230	C ₁₉ H ₁₆ Cl ₂ N ₄ OS		3.85	13.36	780, 1480,
			~		methanol		54.34	4.20	12.95	3220
9a	CO ₂ Et	1	•	80	74 - 75	$C_{16}H_{19}N_3O_3$	63.77	6.36	13.36	1460, 1640,
					diisopropyl ether		63.50	6.64	13.60	1730
9b	CO ₂ Et	0	•	75	137 - 138	C ₁₅ H ₁₇ N ₃ O ₃	62.70	5.97	14.63	1500, 1680,
					ethyl acetate		62.98	6.25	14.94	1720
9c	CN	1	-	80	202 - 203	C ₁₄ H ₁₄ N ₄ O	66.12	5.55	22.03	1480, 1670,
					ethyl acetate	0 11 11 0	66.59	5.54	21.86	2220
9 d	$CH = CH_2$	1	-	80	97 - 96	C ₁₅ H ₁₇ N ₃ O	70.56	6.71	16.46	1500, 1640,
	CH ₃ O				diisopropyl ether	0 11 11 0	70.38	7.04	16.53	1680
9e	NH-C-	1	=	75	185 - 186	$C_{22}H_{24}N_4O_3$	70.19	6.43 6.20	14.88 14.95	1480, 1670, 3250
	— CH₃	4		0.5	ethanol	CHNO	70.32	6.20	14.95	1480, 1680,
9f	Ph	1	-	85	146 - 147 [c] ethyl acetate	C ₁₉ H ₁₉ N ₃ O				2750
9g	p-BrPh	1		75	143 - 144	C ₁₉ H ₁₈ BrN ₃ O	59.38	4.72	10.94	800, 1490,
- 5	F				acetone	,0 .0	59.37	5.06	10.85	1670
10a	-	-	Ph	85	173 - 174	C ₁₉ H ₁₇ N ₃ O ₂	71.45	5.27	13.16	1480, 1600,
					ethyl acetate		72.06	5.82	13.37	1670
10b	-	-	<i>p-</i> BrPh	90	188 - 189	$C_{19}H_{16}BrN_3O_2$	57.30	4.05	10.55	1490, 1620,
					ethyl acetate		57.54	4.35	10.80	1680
10c	-	-	m-ClPh	90	147 - 148	C ₁₉ H ₁₆ CIN ₃ O ₂	64.50	4.56	11.80	1490, 1630,
					methanol		64.10		11.57	1670
10d	-	-	CH ₂ CI	85	161 - 162	C ₁₄ H ₁₄ CIN ₃ O ₂	57.64	4.84	14.40	1490, 1620,
					ethyl acetate		57.41	5.20	14.78	1660
10e [c	i] -	-	CH ₃ - N N - CH ₂ -	65	177 - 180	$C_{19}H_{25}N_5O_2$	64.20		19.71	1480, 1660,
			\bigcup		diisopropyl ether		64.53		19.88	2780
11	-	-	-	95	183 - 185	$C_{18}H_{19}N_3O_5$	60.50		11.76	1130, 1580,
					methanol		60.00		12.05	1680
12	-	-	-	30	181 - 183	$C_{18}H_{23}N_3O_2$	68.98		13.41	1480, 1660,
					ethyl acetate		68.89	7.31	12.94	3400

[a] Prepared from 7a with lithium aluminum hydride. [b] Prepared from 7a with hydrazine hydrate. [c] Lit [3] mp 146-147°.

A mixture of 7a (0.3 g, 1 mmole) and 5 ml of hydrazine hydrate (72%) was dissolved in 10 ml of methanol. After refluxing for 2 hours, the solvent was removed, the crude oily product was dissolved in 40 ml of chloroform, 40 ml of water was then added and the mixture was extracted with a further 2 x 40 ml of chloroform. The combined extract was dried (sodium sulfate) and evaporated to afford a yellow oil, which was crystallized from methanol.

2-Phenylcarbamoyl-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]-pyrimidin-11-one (8a).

A solution of **6** (0.43 g, 2 mmoles) and phenyl isocyanate (0.24 g, 2 mmoles) in 20 ml of benzene was allowed to stand overnight, and the solvent was then removed. The pale-yellow crystalline product **8a** was recrystallized from benzene.

2-Phenylthiocarbamoyl-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido-[1,2-a:4,3-d]pyrimidin-11-one (**8b**).

Compound 8b was obtained as described for 8a, starting from 6 (0.43 g, 2 mmoles) and phenyl isothiocyanate (0.27 g, 2 mmoles).

[[]d] Prepared also from 10d with N-methylpiperazine.

2-(o,o'-Dichlorophenylthiocarbamoyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (**8c**).

This was prepared as described for 8a, starting from 6 (0.43 g, 2 mmoles) and o, o'-dichlorophenyl isothiocyanate (0.41 g, 2 mmoles).

2-(Ethoxycarbonyl-methyl)-9-methyl-1,2,3,4-tetrahydro-11*H*-dipyrido-[1,2-a:4,3-d]pyrimidin-11-one (9a).

Anhydrous potassium carbonate (0.55 g, 4 mmoles) was suspended in a solution of 6 (0.43 g, 2 mmoles) and ethyl bromoacetate (0.33 g, 2 mmoles) in 50 ml of dry acetone. After stirring and refluxing for 5 hours, the suspension was cooled, and the inorganic salt was filtered off and washed with 3 x 5 ml of acetone. The solvent was removed, and the crude oily product 9a was crystallized from di-isopropyl ether.

2-Ethoxycarbonyl-9-methyl-1,2,3,4-tetrahydro-11*H*-dipyrido[1,2-a:4,3-d]-pyrimidin-11-one (**9b**).

This obtained as described for 9a, starting from 6 (0.43 g, 2 mmoles) and ethyl chloroformate (0.22 g, 2 mmoles).

2-Cyanomethyl-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]-pyrimidin-11-one (**9c**).

Compound 9c was obtained as described for 9a, starting from 6 (0.43 g, 2 mmoles) and chloracetonitrile (0.15 g, 2 mmoles).

2-Allyl-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (9d).

This compound was obtained as described for **9a**, starting from **6** (0.43 g, 2 mmoles) and allyl bromide (0.24 g, 2 mmoles).

2-[(o,o'-Dimethyl)-2'-acetanilidyl]-9-methyl-1,2,3,4-tetrahydro-11*H*-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (9e).

Compound **9e** was obtained as described for **9a**, starting from **6** (0.43 g, 2 mmoles) and α -bromo-(0,0'-dimethyl)-acetanilide (0.48 g, 2 mmoles).

2-Benzyl-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (9f).

Compound 9f was obtained as described for 9a, starting from 6 (0.43 g, 2 mmoles) and benzyl bromide (0.34 g, 2 mmoles).

2-(p-Bromobenzyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]-pyrimidin-11-one (9g).

This was obtained as described for 9a, starting from 6 (0.43 g, 2 mmoles) and p-bromobenzyl bromide (0.5 g, 2 g mmoles).

2-Benzoyl-9-methyl-1,2,3,4-tetrahydro-11*H*-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (10a).

To a mixture of $\bf 6$ (0.43 g, 2 mmoles), benzoyl chloride (0.28 g, 2.1 mmoles) and 2 M sodium hydroxide (2 ml, 4 mmoles), 50 ml of benzene was added. The mixture was stirred for 4 hours at ambient temperature. After this the organic layer was separated and dried (sodium sulfate), and the benzene was removed. The off-white crystalline product was recrystallized from ethyl acetate.

 $2{p-Bromobenzoyl}-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]-pyrimidin-11-one (10h).$

This was obtained as described for 10a, starting from 6 (0.43 g, 2 mmoles) and p-bromobenzoyl chloride (0.44 g, 2.1 mmoles).

2-(m-Chlorobenzoyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (**10c**).

This was obtained as described for 10a, starting from 6 (0.43 g, 2 mmoles), and m-chlorobenzoyl chloride (0.35 g, 2.1 mmoles).

2-Chloroacetyl-9-methyl-1,2,3,4-tetrahydro-11*H*-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (**10d**).

Compound 10d was obtained as described for 10a, starting from 6 (0.43 g, 2 mmoles) and chloroacetyl chloride (0.23 g, 2.1 mmoles).

- 2-[(1'-N-Methylpiperazyl)-acetyl]-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (10e).
- i) This compound was obtained as described for **10a**, starting from **6** (0.43 g, 2 mmoles) and (1-N-methylpiperazyl)acetyl chloride (0.35 g, 2.1 mmoles).
- ii) A mixture of 10d (0.58 g, 2 mmoles), N-methylpiperazine (0.2 g, 2 mmoles) and potassium carbonate (0.28 g, 4 mmoles) was suspended in 30 ml of toluene. After refluxing and stirring for 15 hours, the mixture was cooled. The inorganic salt was filtered off, and the product was crystallized from diisopropyl ether, yield, 75%, mp 178-180°.

2- $(\alpha,\beta$ -Dimethoxycarbonylvinyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (11).

A solution of 6 (0.43 g, 2 mmoles) and dimethyl acetylenedicarboxylate (0.28 g, 2 mmoles) in 30 ml of methanol was allowed to stand at ambient temperature for 1 hour. After removal of the solvent, the crude yellow oily product was crystallized from methanol.

2-(trans-2'-Hydroxycyclohexyl)-9-methyl-1,2,3,4-tetrahydro-11H-dipyrido[1,2-a:4,3-d]pyrimidin-11-one (12).

A mixture of 6 (0.43 g, 2 mmoles) and cyclohexene oxide (0.39 g, 4 mmoles) was dissolved in 20 ml of methanol. After refluxing for 25 hours, the solvent was removed. To the crude oily residue, 30 ml of water was added. This was extracted with 3 x 50 ml of benzene. The combined extract was dried (sodium sulfate) and the benzene was removed. The crude yellow oily product was crystallized from ethyl acetate.

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